
Synthesis and Biological activity of Some PyrroloPyrimidine Compounds.

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Abstract

The present work was planned to synthesize some heterocyclic compounds containing pyrrolo[2,3-d]pyrimidines.

Our approach to the synthesis of the target compounds started through the hydrolysis of 6-ethylthio-4-methyl-2-(p-methoxyphenyl)pyrimidine-5-carbonitrile by refluxing in ethanol containing sodium hydroxide to give the corresponding sodium salt derivative which in turn was acidified to give 5-cyano-4-methyl-2-(p-methoxyphenyl)pyrimidin-6(H)one. Chlorination of the 5-Cyano-4-methyl-2-(p-methoxyphenyl)pyrimidin-6(1H)-one afforded 6-Chloro-4-methyl-2-(p-methoxyphenyl)pyrimidin-5-carbonitrile, which reacted with ethyl glycinate hydrochloride to give Ethyl (5-cyano-4-methyl-2-(p-methoxyphenyl)pyrimidin-6-yl)acetate, which underwent cyclization to give the ethyl 5-amino-4-methyl-2-(p-methoxyphenyl)-7H-pyrrolo[2,3-d]pyrimidine-6-carboxylate. The reaction of 5-Cyano-6-chloro-4-methyl-2-(p-methoxyphenyl)pyrimidine with piperidine, and morpholine, gave the 4-Methyl-6-(morpholin-4-yl)-2-(p-methoxyphenyl)pyrimidine-carbonitrile and 4-Methyl-6-(morpholin-4-yl)-2-(p-methoxyphenyl)pyrimidine-5-carbonitrile respectively.

The Ethyl 5-amino-4-methyl-2-(p-methoxyphenyl)-7H-pyrrolo[2,3-d]pyrimidine-6-carboxylate was used as key intermediates in the synthesis of other pyrrolo[2,3-d]pyrimidine derivatives, pyrimido[4,5:4,5]pyrrolo[2,3-d]pyrimidines and triazolopyrrolopyrimidines by reaction with different reagents.

Seven compounds were selected and screened *in vitro* for their antimicrobial activity against four strains of bacteria and two fungal species.

Key words : pyrrolopyrimidines, Pyrimidopyrimidines, synthesis , Biological activity

Introduction:

Five positional isomers of pyrrolopyrimidines are known as a result of fusion of pyrrole to the pyrimidine nucleus: pyrrolo[2,3-d]pyrimidine, pyrrolo[3,2-d]pyrimidine, pyrrolo[3,4-d]pyrimidine, pyrrolo[1,2-a]pyrimidine and pyrrolo[1,2-c]pyrimidine. Two synthetic routes have been used for the synthesis of pyrrolo[2,3-d]pyrimidine either starting with the pyrrole moiety or the pyrimidine one. The literature survey reveals that the most important one is

that starting from the appropriate ortho-substituted aminopyrroles. Pyrimidine derivatives have occupied a unique position in medicinal chemistry, the pyrimidine ring is present in a large number of biological important compounds such as alkaloids. Condensed pyrimidines have received much attention over the years because of their interesting biological and medicinal importance such as antiviral [1], antitumors [2, 3], anti cancer [4, 5], competitive inhibitors

[6, 7], antimicrobial [8, 9], antiparasitic activities [10] as well as antiproliferative activity [11]. In view of the above observation and as a continuation of our previous work about the synthesis of fused

Experimental:

All melting points are uncorrected and measured on a Gallan-Kamp apparatus. IR spectra were recorded on a Shimadzu-470 IR-spectrophotometer (KBr; ν_{\max} in cm^{-1}). ^1H NMR spectra were measured on a Varian EM-390, 90 MHz spectrometer or on a Jeol LA 400 MHz FT-NMR spectrometer with TMS as internal standard (δ in ppm); MS on a Jeol JMS-600 mass spectrometer. Elemental analyses were determined on a Perkin-Elmer 240C elemental analyzer or on an Elemental Analyses system GmbH VARIOEL V_{2,3} CHNS Mode.

5-Cyano-4-methyl-2-(p-methoxyphenyl)pyrimidin-6(1H)-thione (1):

Prepared as previously reported in literature procedure [14].

The synthesized compounds (2-19) were prepared with some modification of reported methods in literature procedures [12-14].

6-Ethylthio-4-methyl-2-(p-methoxyphenyl)pyrimidin-5-carbonitrile (2):

A mixture of **1** (2.57 g, 0.01 mol.) and ethyl bromide (0.015 mol) in ethanol (50 ml) was refluxed for 3 hrs., in the presence of anhydrous potassium carbonate (0.5g). The solid product separated from the hot mixture was filtered off, washed with water and recrystallized from ethanol as pale yellow crystals.

Sodium 5-cyano-4-methyl-2-(p-methoxyphenyl)-6-pyrimidinate (3):

A mixture of **2** (2.85 g, 0.01 mol.), sodium hydroxide (0.02 mol) in water (1 ml) and ethanol (20 ml), was refluxed for 4 hrs, then allowed to cool. The solid product was filtered off, washed with ethanol as white crystals.

5-Cyano-4-methyl-2-(p-methoxyphenyl)pyrimidin-6(1H)-one (4):

pyrimidine compounds [12-14], which are expected to be of biological and medicinal importance. We became interested in the synthesis of some heterocyclic compounds containing pyrrolo[2,3-d]pyrimidines.

To a solution of **3** (2.63 g, 0.01 mol.) in water, 50 ml of HCl was added. The white precipitate which separated was collected by filtration and air dried. Recrystallization from DMF gave white crystals.

5-Cyano-6-chloro-4-methyl-2-(p-methoxyphenyl)pyrimidine (5):

A sample of **3** (2.63 g, 0.01 mol.) in phosphorus oxychloride (15 ml) was heated under reflux for 4 hrs. After cooling the reaction mixture was poured into ice/water mixture (200g) drop wise with stirring. The solid product was filtered off and recrystallized from ethanol as pale yellow crystals.

Ethyl (5-cyano-4-methyl-2-(p-methoxyphenyl)pyrimidin-6-yloxy)acetate (6):

A mixture of **3** (2.63 g, 0.01 mol.) and ethyl chloroacetate in pure acetone (20 ml) was refluxed for 6 hrs, then allowed to cool and poured into cold water with stirring. The solid product was collected and recrystallized from ethanol as white crystals.

4-Methyl-2-(p-methoxyphenyl)-6-(piperidin-1-yl)pyrimidine-5-carbonitrile (7):

A sample of compound **5** (1.30 g, 0.005 mol.) in piperidine (2 ml) was fused for 15 min, then ethanol (20 ml) was added and the mixture was refluxed for 2 hrs. The solid product obtained after cooling was filtered off and recrystallized from ethanol as pale yellow crystals.

4-Methyl-6-(morpholin-4-yl)-2-(p-methoxyphenyl)pyrimidine-5-carbonitrile (8):

A sample of compound **5** (1.3 g, 0.005 mol.) and morpholine (2 ml) was fused for 15 min, then ethanol (20 ml) was added and the mixture was refluxed for 2 hrs. The solid product obtained after cooling was filtered off and recrystallized from ethanol as pale yellow crystals.

Ethyl N-(5-cyano-4-methyl-2-(p-methoxyphenyl)pyrimidin-6-yl)glycinate (9):

A mixture of **5** (2.6 g, 0.01 mol.), ethyl glycinate hydrochloride (2.1 g, 0.015 mol) and K₂CO₃ (0.02 mol.) in DMF (30 ml) was heated on steam bath at (70-80°C) for 8 hrs. After cooling the reaction mixture was poured into ice-water mixture (50 gm) slowly with stirring. The solid product was filtered off and recrystallized from ethanol as white crystals.

Ethyl 5-amino-4-methyl-2-(p-methoxyphenyl)-7H-pyrrolo[2,3-d]pyrimidine-6-carboxylate (10):

A sample of **9** (1.63 g, 0.005 mol.) in absolute ethanol (50 ml) containing sodium metal (0.5 g) was refluxed for 1 hr. The solid product was collected by filtration, washed thoroughly with water and recrystallized from ethanol as yellow crystals.

Sodium 5-amino-4-methyl-2-(p-methoxyphenyl)-7H-pyrrolo[2,3-d]pyrimidine-6-carboxylate (11):

A mixture of **10** (3.26 g, 0.01 mol.) in ethanolic sodium hydroxide solution (0.6g) in ethanol (30 ml) was heated under reflux for 5 hrs. The solid product was collected, washed with ethanol and air dried. The yellowish white solid was subjected to the next step without further purification.

4-Methyl-2-(p-methoxyphenyl)-5,6-dihydro-7H-pyrrolo[2,3-d]pyrimidin-5-one (12):

A sample of **11** (1.6 g, 0.005 mol.) in orthophosphoric acid (10 ml) was stirred at room temperature for 3 hrs then poured into ice-water mixture (200 g) and neutralized by ammonium hydroxide. The white precipitate was collected and recrystallized from ethanol as white crystals.

5-Amino-4-methyl-2-(p-methoxyphenyl)-7H-pyrrolo[2,3-d]pyrimidine-6-carbohydrazide (13):

A mixture of **10** (1.63 g, 0.005 mol.) and hydrazine hydrate 99% (3 ml) in ethanol (50 ml) was refluxed for 2 hrs. The solid product which formed on hot was filtered off and recrystallized from dioxan as yellow crystals.

9-Methyl-7-(p-methoxyphenyl)pyrimido[4,5[\]:4,5]-5H-pyrrolo[2,3-d]pyrimidin-4(3H)-one (14):

A sample of **10** (1.63 g, 0.005 mol.) in formamide (15 ml) was refluxed for 3 hrs. The solid product which formed on hot was filtered off and recrystallized from dioxan as yellow crystals.

4-Chloro-9-methyl-7-(p-methoxyphenyl)-5H-pyrimido[4,5[\]:4,5]pyrrolo[2,3-d]pyrimidine (15):

A sample of **14** (3.07 g, 0.01 mol.) in phosphorus oxychloride (15 ml) was heated under reflux for 4 hrs. After cooling the reaction mixture was poured into ice/water mixture dropwise with stirring. The solid product was filtered off and recrystallized from ethanol as pale yellow crystals.

4-Hydrazino-9-methyl-7-(p-methoxyphenyl)-5H-pyrimido[4,5[\]:4,5]pyrrolo[2,3-d]pyrimidine (16):

A mixture of **15** (3.25 g, 0.01 mol.) and hydrazine hydrate 99% (0.015 mol.) in ethanol (50 ml) was refluxed for 3 hrs. The solid product which separated from the hot mixture was filtered off and recrystallized from dioxan as pale yellow crystals.

9-Methyl-4-(p-substitutedbenzylidenehydrazono)-7-(p-methoxyphenyl)-5H-pyrimido[4,5[\]:4,5]pyrrolo[2,3-d]pyrimidine (17a-c):

General procedure:

A mixture of **16** (1.60 g, 0.005 mol.) and appropriate aromatic aldehyde (0.005 mol.) in 10 ml acetic acid was refluxed for 3 hrs, then allowed to cool. The solid product was collected and recrystallized from acetic acid as yellow crystals.

9-Methyl-7-(p-methoxyphenyl)-5H-[1,2,4]triazolo[4,3:1,6]pyrimido[4,5[\]:4,5]pyrrolo[2,3-d]pyrimidine (18):

A mixture of hydrazine derivative **16** (0.01 mol) and triethyl orthoformate (2 ml) in ethanol (30 ml) in the presence of a few drops of acetic acid was refluxed for 3 hrs. The solid crystals separated from the hot mixture

were filtered off and recrystallized from dioxane as yellow crystals .

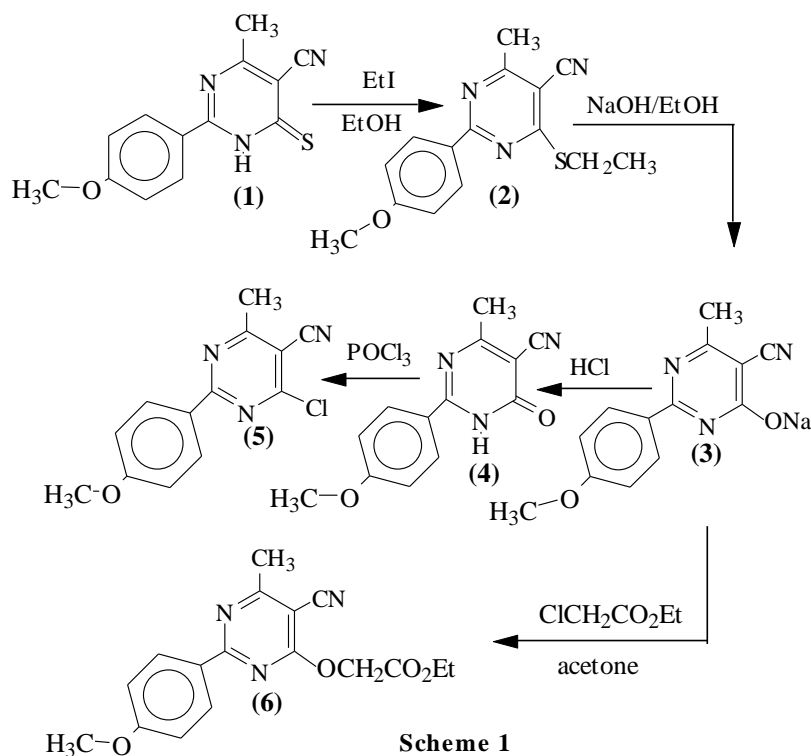
9-Methyl-7-(p-methoxyphenyl)-5H-2-methyl-[1,2,4]triazolo[4,3:1,6]pyrimido[4,5 : 4,5]pyrrolo[2,3-d]pyrimidine (19):

Results and discussion :

The approach to the synthesis of the target compounds started from the hydrolysis of 6-ethylthio-4-methyl-2-(4-methoxyphenyl) pyrimidine-5-carbonitrile (2) by refluxing in ethanol containing sodium hydroxide to give the corresponding sodium salt derivative (3) which in turn was acidified to give the

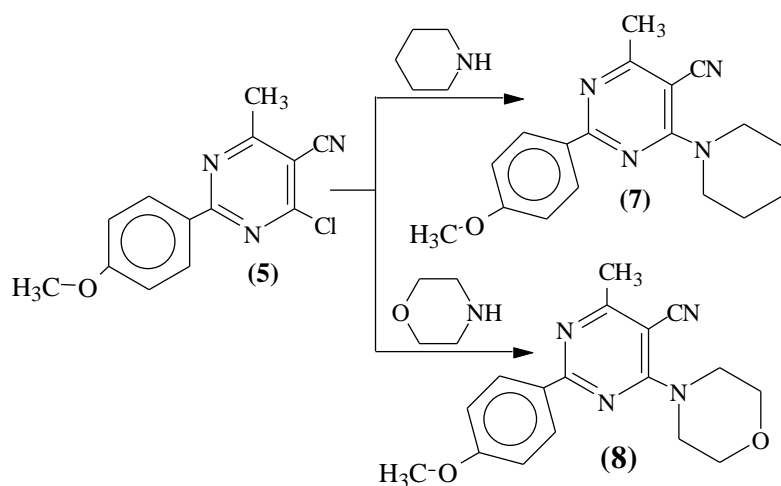
A mixture of hydrazine derivative 16 (0.01 mol) and acetic anhydride (20 ml) was refluxed for 4 hrs. The solid crystals separated from the hot mixture were filtered off and recrystallized from DMF as yellow crystals .

hydroxy derivative (4). Chlorination of (4) with phosphorus oxychloride afforded 6-chloropyrimidine derivative (5) . Also the reaction of (3) with ethyl chloroacetate, by refluxing in ethanol or acetone gave the corresponding O-substituted oxopyrimidine (6) (Scheme 1).



On treatment of compound (5) with piperidine, and morpholine, nucleophilic displacement took place to give the

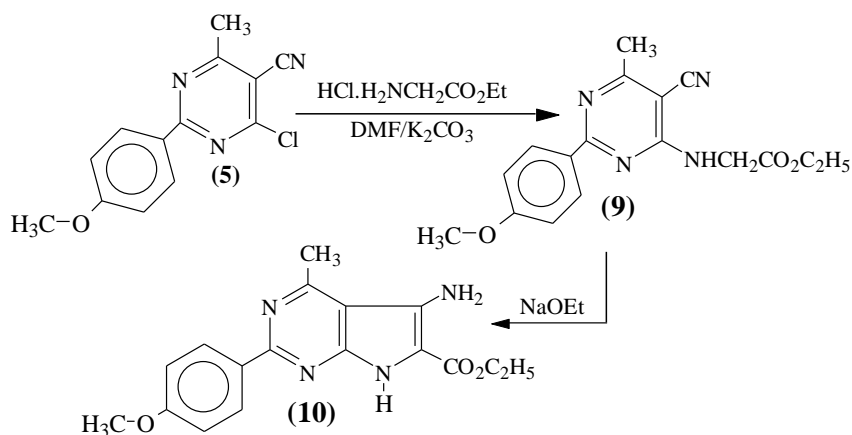
piperidino- and morpholino- derivatives (7) and (8) respectively (Scheme 2).



Scheme 2

6-Chloro-4-methyl-2-(p-methoxyphenyl)pyrimidine-5-carbonitrile (**5**) was employed as starting material for the target compound pyrrolopyrimidine. The reaction of compound (**5**) with ethyl glycinate hydrochloride by heating in DMF in the presence of anhydrous K_2CO_3 gave the corresponding glycinate

derivative (**9**) which underwent intramolecular *Thorpe-Ziegler* cyclization upon treatment with sodium ethoxide to give ethyl 5-amino-4-methyl-2-(p-methoxyphenyl)-7H-pyrrolo[2,3-d]pyrimidin-6-carboxylate (**10**) (Scheme 3).



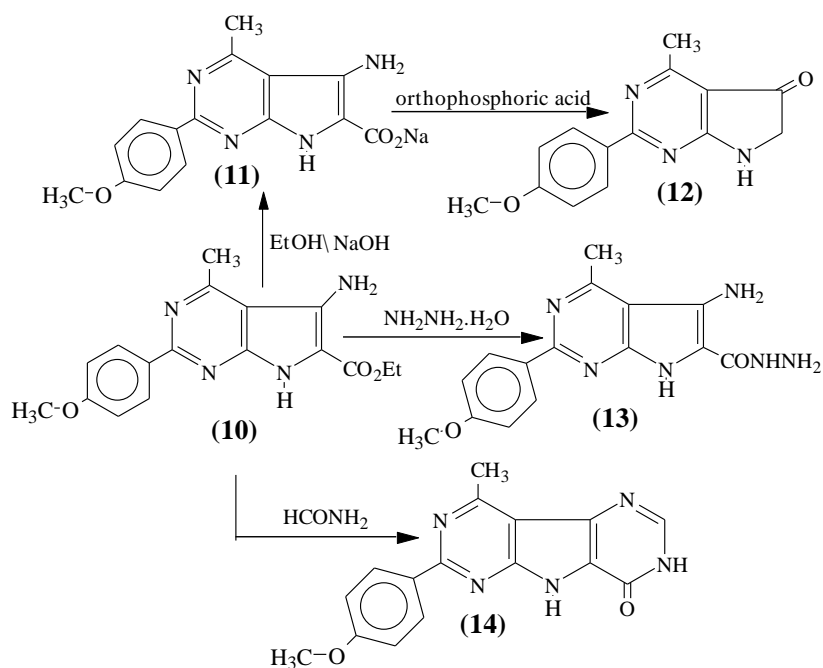
Scheme 3

The *o*-amino ester (**10**) was used as a key intermediate in the synthesis of other new pyrrolo[2,3-d]pyrimidines and pyrimido[4',5':4,5]pyrrolo[2,3-d]pyrimidine (Scheme 3).

Thus saponification of compound (**10**) with an ethanolic sodium hydroxide solution afforded (**11**) which on treatment with orthophosphoric acid gave 4-methyl-2-(4-methoxyphenyl)-5,6-dihydro-7H-pyrrolo[2,3-d]pyrimidin-5-one (**12**).

The reaction of compound (**10**) with an excess amount of hydrazine hydrate (99%) in refluxing ethanol produced 5-amino-4-methyl-2-(4-methoxyphenyl)-7H-pyrrolo[2,3-d]pyrimidine-6-carbohydrazide (**13**)

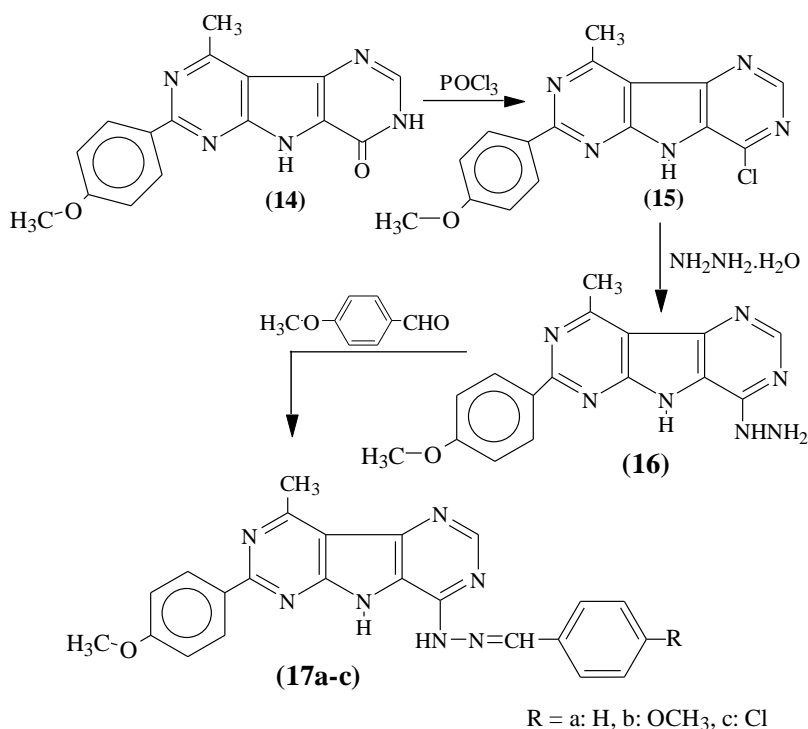
Also compound (**10**) underwent ring closure reaction upon treatment with formamide to afford the 9-Methyl-7-(4-methoxyphenyl)pyrimido[4',5':4,5]-5H-pyrrolo[2,3-d]pyrimidin-4(3H)-one (**14**) (Scheme 4).



Scheme 4

phenyl)-5H-pyrimido[4[\],5[\]:4,5]pyrrolo[2,3-d]pyrimidine (16). The latter compound was used as precursor for synthesizing 9-Methyl-4-(4-methoxybenzylidenehydrazono)-7-(4-methoxyphenyl)-5H-pyrimido[4[\],5[\]:4,5]pyrrolo[2,3-d]pyrimidine (17a-c) by reaction with a variety of aromatic aldehydes in acetic acid (Scheme 5).

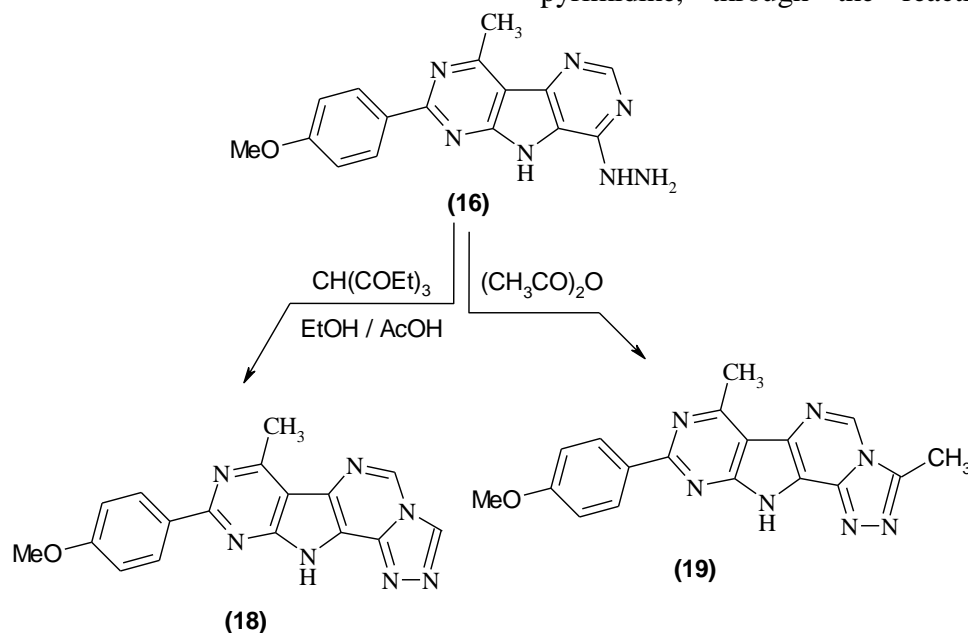
Chlorination of compound (14) by refluxing with an excess amount of phosphorus oxychloride gave 4-Chloro-9-methyl-7-(4-methoxyphenyl)-5H-pyrimido[4[\],5[\]:4,5]pyrrolo[2,3-d]pyrimidine (15), which reacted with an equimolar amount of hydrazine hydrate (99%) in refluxing ethanol produced 4-hydrazino-9-methyl-7-(4-methoxyphenyl)-5H-pyrimido[4[\],5[\]:4,5]pyrrolo[2,3-d]pyrimidine (16).



Scheme 5

reagents such as triethyl orthoformate and acetic anhydride to give compounds (18) and (19) respectively (Scheme 6).

The hydrazino derivative (16) was used as a key intermediate to synthesize a new ring system, namely triazolopyrrolo pyrimidine, through the reaction with



Scheme 6

The structural formulae of the synthesized compounds (2-19) were confirmed by elemental and spectral analyses as shown in tables (2&3).

Biological Activity :

Seven of the synthesized compounds have been tested against two pathogenic gram positive bacteria, *Bacillus cereus* and *Staphylococcus aureus* and two gram negative bacteria : *Serratia marcescens* and *Pseudomonas aeruginosa*. Also three fungal

species were used in the present investigation : *Aspergillus*, *Penicillium chrysogenum* and *Fusarium moniliforme*. The biological activity as expressed by the growth inhibition zone of the tested microorganism listed in table 1. It is obvious that some of the tested compounds exhibit the moderate activity (inhibition zone 6-10 mm) against the mentioned species of bacteria and fungi as shown in table 1.

Table 1: Results of the antimicrobial screening

Comp.No	B. cereus	S. aureus	S. marcescens	P. aeruginosa	A. terreus	P. chrysogenum	F. moniliforme
5	10	8	-	-	7	8	8
8	-	-	-	-	-	-	-
10	10	8	-	-	7	8	8
11	8	-	-	-	-	-	-
13	10	8	-	-	8	-	10
16	10	8	-	-	8	8	-
17a	8	6	-	-	7	6	-

Table2: Physical Properties and Analytical Data of Compounds (2-19)

.No	m.p ^o C	Yield %	Formula/mol.wt	Calculated / Found		
				C	H	N
2	125-127	70	C ₁₅ H ₁₅ N ₃ OS 285.37	63.13	5.30	14.72
				63.15	5.27	14.70
3	-----	61	C ₁₃ H ₁₀ N ₃ O ₂ Na 263.20	59.32	3.83	15.96
				59.18	3.76	16.02
4	285-287	76	C ₁₃ H ₁₁ N ₃ O ₂ 241.25	64.72	4.60	17.42
				65.01	4.58	17.33
5	184-186	71	C ₁₃ H ₁₀ ClN ₃ O 259.70	60.13	3.88	16.18
				60.07	3.90	16.30
6	123-125	73	C ₁₇ H ₁₇ N ₃ O ₄ 327.34	62.38	5.23	12.84
				62.26	5.19	13.01
7	102-104	68	C ₁₈ H ₂₀ N ₄ O 308.39	70.11	6.54	18.17
				70.32	6.50	18.09
8	175-177	90	C ₁₇ H ₁₈ N ₄ O ₂ 310.36	65.79	5.85	18.05
				65.70	6.00	18.10
9	142-144	90	C ₁₇ H ₁₈ N ₄ O ₃ 326.36	62.57	5.56	17.17
				62.58	5.54	17.21
10	163-165	75	C ₁₇ H ₁₈ N ₄ O ₃ 326.36	62.57	5.56	17.17
				62.60	5.51	17.25
12	143-145	65	C ₁₄ H ₁₃ N ₃ O ₂ 255.28	65.87	5.13	16.46
				66.01	5.08	16.37
13	310-315	68	C ₁₅ H ₁₆ N ₆ O ₂ 312.33	57.68	5.16	26.91
				57.60	5.10	27.03
14	>350 decompose	76	C ₁₆ H ₁₃ N ₅ O ₂ 307.31	62.53	4.26	22.79
				62.45	4.35	22.82
15	315-317	69	C ₁₆ H ₁₂ ClN ₅ O 325.76	58.99	3.71	21.50
				59.05	3.68	21.51
16	285-287	65	C ₁₆ H ₁₅ N ₇ O 321.34	59.80	4.71	30.51
				60.00	4.65	30.49
17a	295-297	68	C ₂₃ H ₁₉ N ₇ O 409.44	67.47	4.68	23.95
				67.58	4.65	24.06
17b	310-312	69	C ₂₄ H ₂₁ N ₇ O ₂ 439.48	65.59	4.82	22.31
				65.60	4.90	22.20
17c	>300 decompose	72	C ₂₃ H ₁₈ ClN ₇ O 443.90	62.23	4.09	22.09
				62.15	4.00	22.10
18	>320 decompose	75	C ₁₇ H ₁₃ N ₇ O 331.34	61.63	3.95	29.59
				61.55	4.00	29.54
19	>320 decompose	63	C ₁₈ H ₁₅ N ₇ O 345.37	62.60	4.38	28.39
				62.75	4.31	28.42

5 Cl (Cal. 13.65, Found 13.50).

15 Cl (Cal. 10.88, Found 11.00).

17c Cl (Cal. 7.99, Found 8.05).

Table 3 Spectroscopic Data of Compounds (2-17)

2	IR: $\nu = 2200\text{cm}^{-1}$ (C \equiv N). $^1\text{HNMR}$ (DMSO- d_6) : $\delta = 1.3\text{-}1.5$ (t, 3H, CH $_3$), 2.8 (s,3H, CH $_3$, pyrimidine), 3.3-3.5 (q, 2H, CH $_2$), 3.9 (s, 3H, OCH $_3$) 7.3, 7.4 (d, 2H, Ar-H) and 7.9, 8.0 (d, 2H, Ar-H).
4	IR: $\nu = \text{IR: } \nu = 3450\text{ cm}^{-1}$ (NH), 2200 cm^{-1} (C \equiv N) and 1680cm^{-1} (C=O). $m/z = 241$
5	IR: $\nu = 2200\text{cm}^{-1}$ (C \equiv N) and no bands characteristic for (NH, C=O) . $^1\text{HNMR}$ (DMSO- d_6) : $\delta = 2.8$ (s,3H, CH $_3$, pyrimidine), 3.85 (s, 3H, OCH $_3$), 7.3, 7.4 (d, 2H, Ar-H) and 7.9, 8.0 (d, 2H, Ar-H). $m/z = 259$ (98%), 261(31.8%), 224(M $^+$ -Cl).
6	IR: $\nu = \text{IR: } \nu = 2200\text{ cm}^{-1}$ (C \equiv N) and 1730 cm^{-1} (C=O). $^1\text{HNMR}$ (CDCl $_3$) : $\delta = 1.15\text{-}1.3$ (t, 3H, CH $_3$, ester), 2.8 (s,3H, CH $_3$, pyrimidine), 3.85 (s, 3H, OCH $_3$), 4.05-4.30 (q, 2H, CH $_2$, ester), 5.0 (s, 2H, CH $_2$), 7.3, 7.4 (d, 2H, Ar-H) and 7.9, 8.0 (d, 2H, Ar-H).
7	IR: $\nu = \text{IR: } \nu = 2200\text{ cm}^{-1}$ (C \equiv N) . $^1\text{HNMR}$ (DMSO- d_6) : $\delta = 1.6$ (s,6H, 3CH $_2$), 2.6 (s,3H, CH $_3$, pyrimidine), 3.2(s,4H, CH $_2$ -N-CH $_2$), 3.85 (s, 3H, OCH $_3$),7.3, 7.4 (d, 2H, Ar-H) and 7.9, 8.0 (d, 2H, Ar-H).
8	IR: $\nu = \text{IR: } \nu = 2200\text{ cm}^{-1}$ (C \equiv N) . $^1\text{HNMR}$ (DMSO- d_6) : $\delta = 2.7$ (s,3H, CH $_3$, pyrimidine), 3.4 (s, 4H, CH $_2$ -N-CH $_2$), 3.9 (s, 4H, CH $_2$ -O-CH $_2$), 4.1 (s, 3H, OCH $_3$), 7.3, 7.4 (d, 2H, Ar-H) and 7.9, 8.0 (d, 2H, Ar-H).
9	IR: $\nu = 3400\text{ cm}^{-1}$ (NH), 2200 cm^{-1} (C \equiv N) and 1700 cm^{-1} (C=O). $^1\text{HNMR}$ (CDCl $_3$) : $\delta = 1.3\text{-}1.5$ (t, 3H, CH $_3$, ester), 2.7 (s, 3H, CH $_3$, pyrimidine), 4.1 (s, 3H, OCH $_3$), 4.2- 4.5 (m, 4H, 2CH $_2$), 6.0 (s,1H, NH), 7.3, 7.4 (d, 2H, Ar-H) and 7.9, 8.0 (d, 2H, Ar-H).
10	IR: $\nu = 3450, 3350\text{ cm}^{-1}$ (NH $_2$), 3100 cm^{-1} (NH) and 1680 cm^{-1} (C=O). $^1\text{HNMR}$ (CDCl $_3$) : $\delta = 1.9\text{-}1.32$ (J = 7.0 Hz, t, 3H, CH $_3$, ester), 2.87 (s,3H, CH $_3$, pyrimidine), 4.1 (s, 3H, OCH $_3$), 4.8 (broad signal, 2H, NH $_2$), 4.25-4.31 (J = 7.0 Hz, q, 2H, CH $_2$, ester), 7.3, 7.4 (d, 2H, Ar-H), 7.9, 8.0 (d, 2H, Ar-H and 11.0 (s,1H, NH).
12	IR: $\nu = 3250\text{ cm}^{-1}$ (NH) and 1730 cm^{-1} (C=O). $^1\text{HNMR}$ (DMSO- d_6) : $\delta = 2.6$ (s, 3H, CH $_3$, pyrimidine), 4.1 (s, 3H, OCH $_3$), 4.5, 4.6 (d, 2H, CH $_2$), 6.5 (broad signal,1H, NH), 7.3, 7.4 (d, 2H, Ar-H) and 7.9, 8.0 (d, 2H, Ar-H).
13	IR: $\nu = 3400\text{-}3200\text{ cm}^{-1}$ (NH, NH, NH $_2$, NH $_2$) and 1650 cm^{-1} (C=O) . $^1\text{HNMR}$ (DMSO- d_6) : $\delta = 2.81$ (s,3H, CH $_3$, pyrimidine), 4.44 (s. 2H, NH $_2$, hydrazide group), 4.1 (s, 3H, OCH $_3$), 5.82 (s. 2H, NH $_2$, pyrrole), 7.3, 7.4 (d, 2H, Ar-H) and 7.9, 8.0 (d, 2H, Ar-H , 8.79 (s, 1H, NH) and 11.40 (s. 1H, NH, pyrrole),
14	IR: $\nu = 3150\text{-}3400\text{ cm}^{-1}$ (2NH) and 1680 cm^{-1} (C=O) . $^1\text{HNMR}$ (DMSO- d_6) : $\delta = 3.0$ (s, 3H, CH $_3$, pyrimidine), 4.1 (s, 3H, OCH $_3$), 7.3, 7.4 (d, 2H, Ar-H), 7.9, 8.0 (d, 2H, Ar-H), 8.11 (s, 1H, CH), 9.83(s, 1H, NH), and 12.92 (s. 1H, NH, pyrrole).
15	IR: $\nu = 3100\text{ cm}^{-1}$ (NH) . Mass spectrum showed M $^+$ and M+2 peaks at $m/z = 325$ (100%), 327 (32.7%).
16	IR: $\nu = 3400\text{-}3300\text{ cm}^{-1}$ (NH, NH, NH $_2$, NH $_2$). $^1\text{HNMR}$ (DMSO- d_6) : $\delta = 3.0$ (s,3H, CH $_3$, pyrimidine), 3.54 (s,2H, NH $_2$, hydrazino), 4.1 (s, 3H, OCH $_3$), 7.3, 7.4 (d, 2H, Ar-H), 7.9, 8.0 (d, 2H, Ar-H), 8.13(s,1H,NH, hydrazino), 8.45(s,1H,CH),10.13 (s,1H, NH, pyrrole) .

17a	IR: $\nu = 3200-3000 \text{ cm}^{-1}$ (2NH). $^1\text{HNMR}$ (DMSO- d_6) : $\delta = 3.0$ (s, 3H, CH ₃ , pyrimidine), 3.9 (s, 3H, OCH ₃), 7.01 (s,1H, HN-N=C), 8.18 (s, 1H, HC=N), 7.51-8.55 (m,9H, Ar-H), 8.59 (s,1H, CH, pyrimidine) and 9.84 (s, 1H, NH, pyrrole) .
17b	IR: $\nu = 3200-3000 \text{ cm}^{-1}$ (2NH). $^1\text{HNMR}$ (DMSO- d_6) : $\delta = 3.0$ (s, 3H, CH ₃ , pyrimidine), 3.9 (s, 6H, 2 OCH ₃), 7.01 (s,1H, HN-N=C), 8.18 (s, 1H, HC=N), 7.40-8.35 (m,8H, Ar-H), 8.59 (s,1H, CH, pyrimidine) and 9.84 (s, 1H, NH, pyrrole) .
17c	IR: $\nu = 3200-3000 \text{ cm}^{-1}$ (2NH). $^1\text{HNMR}$ (CF ₃ COOD) : $\delta = 3.0$ (s, 3H, CH ₃ , pyrimidine), 4.1 (s, 3H, OCH ₃), 8.3 (s, 1H, HC=N), 7.40-8.4 (m,8H, Ar-H) and 8.6 (s,1H, CH, pyrimidine) .
18	IR: $\nu = 2950 \text{ cm}^{-1}$ (CH aliph.), and 1600 cm^{-1} (C=N). $^1\text{HNMR}$ (CF ₃ COOD) : $\delta = 2.9$ (s,3H, CH ₃ , pyrimidine), 4.1 (s, 3H, OCH ₃), 7.3, 7.4 (d, 2H, Ar-H), 7.9, 8.0 (d, 2H, Ar-H) and 8.6, 9.1(2s,2H,2CH triazole, pyrimidine),.
19	IR: $\nu = 2950 \text{ cm}^{-1}$ (CH aliph.), and 1620 cm^{-1} (C=N). $^1\text{HNMR}$ (CF ₃ COOD) : $\delta = 2.9$ (s,6H, 2CH ₃ , pyrimidine and triazole), 4.1 (s, 3H, OCH ₃), 7.3, 7.4 (d, 2H, Ar-H), 7.9, 8.0 (d, 2H, Ar-H) and 9.2 (s, H, CH, pyrimidine),.

Low ester carbonyl stretching frequencies around 1680 cm^{-1} was found in the IR spectrum as a result of intramolecular hydrogen bonding with the ortho-amino group [15].

confirmation of their structural formulae on the basis of elemental and spectral analyses . The synthesized compounds posses biological activity against some species of bacteria and fungi .

Conclusion :

Synthesis of some heterocyclic compounds related to pyrimidine moiety and

تخليق وتفاعلات بعض مركبات بيرولوبيريميدين.

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الملخص

أستهدف هذا البحث تحضير بعض المركبات الحلقية غير المتجانسة الجديدة المتضمنة بيرولوبيريميدين [2،3- د]بيروبيديينات التي من المتوقع أن يكون لها أهمية بيولوجية وطبية. وقد تضمن هذا البحث ايضاً تفاعل هذه المركبات مع بعض الكواشف لتعطى أنظمه غير متجانسة الحلقة ملتحمة معها.

طريقنا لتحضير المركبات المستهدفة بدء من التحلل المائي للمركب 6- ايثيل ثيو - 4- ميثيل - 2-بارا- فينيل بيروبيديين -5- كربونتريل بغليانه في الايثانول المحتوي على هيدروكسيد الصوديوم ليعطي مشتق الملح الصوديومي المقابل والذي تم تحميضه ليعطي مشتق الهيدروكسيل المقابل . كلورة المركب 5-سيانو-4- ميثيل -2- (بارا ميثوكسي فينيل)بيروبيديين-6(ايدون) بواسطة أكسي كلوريد الفوسفور أعطى المشتق 4-

كلوروبيريبيدين. تفاعل الملح الصديومي مع أسيتات كلوريد الايثيل أعطى المركب ايثايل (5-سيانو-4-ميثيل-2- (بارا ميثوكسي فينيل)بيريبيدين-6-يل اوكسي)اسيتيت (6). معالجة المشتق 4- كلوروبيريبيدين بالبيريبيدين و المورفولين أعطى المركبان 4-ميثيل-2- (بارا ميثوكسي فينيل)-6- (بايبيريدين-1-يل)بيريبيدين-5-كربونتريل و 4-ميثيل-6- (مورفولين-4-يل)-2- (بارا ميثوكسي فينيل)-بيريبيدين-5-كربونتريل على التوالي .

تفاعل المركب 5-سيانو-6-كلورو-4-ميثيل-2- (بارا ميثوكسي فينيل)بيريبيدين مع هيدروكلوريد ايثيل جلايسينات أعطى مشتق الجلايسينات المقابلة الذي خضع لعملية تحوّل ليعطي المركب ايثيل 5-أمينو-4-ميثيل-2-بارا-فينيل-7-يد-بيرولو[2،3-د]بيريبيدين-6-كربوكسيلات . وهذا الأخير استخدم في تحضير بعض مشتقات بيرولو[2،3-د]بيريبيدينات ، بيريبيدينو[4،5/4،5]بيرولو[2،3-د]بيريبيدينات و ترازولوبيريبيدينوبيريبيدينات بتفاعله مع كواشف مختلفة . تم دراسة عدد سبعة مركبات ضد أنواع مختلفة من البكتريا والميكروبات .

خلاصة هذا البحث تتمثل في تحضير عدد من المركبات الحلقية الغير متجانسة المنتمية إلى نواة البيرولو[2،3-د]بيريبيدين واثبات تركيبها بأطياف الأشعة تحت الحمراء ، الرنين النووي المغناطيسي والتحليل الكمي للعناصر، بالإضافة إلى معرفة تأثير بعضها على أنواع مختلفة من البكتريا والميكروبات .

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